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(54) Process for the Production of Polyisocyanates Containing
Isocyanurate Groups and Their Use

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A PROCESS FOR THE PRODUCTION OF POLYISOCYANATES
CONTAINING ISOCYANURATE GROUPS AND THEIR USE

BACKGROUND OF THE INVENTION

Field of the Invention

5 The present invention is directed to a new
process for the production of polyisocyanates containing
isocyanurate groups by trimerizing a portion of the
isocyanate groups of hexamethylene diisocyanate
(referred to hereinafter as "HDI"), to the products
10 obtained by this process, and their use, optionally in
blocked form, as the isocyanate component in
polyisocyanate polyaddition products, preferably
polyurethane lacquers.

Description of the Prior Art

15 The use of quaternary ammonium hydroxides as
catalysts for the trimerization of isocyanate groups is
known and has been repeatedly described. Thus,
according to JP-PS 601,337 (US-PS 3,487,080), quaternary
ammonium hydroxides are used together with certain
20 co-catalysts. The examples primarily describe the
partial trimerization of aromatic diisocyanates.
However, the partial trimerization of HDI is described
in the examples.

25 The process according to EP-A 10,589 represents
a further development of the process according to the
Japanese patent specification cited above. According to
this prior publication, quaternary ammonium hydroxides
containing hydroxyalkyl substituents are used for the
trimerization of HDI. With these catalysts, HDI can be
30 trimerized in excellent fashion without cloudiness. The
disadvantage of this process is that the hydroxyalkyl
ammonium hydroxides are very difficult to produce in
colorless form and have to be used in relatively large



quantities of up to 0.6%. Accordingly, the end products of the process, i.e. the isocyanurate polyisocyanates freed from excess starting diisocyanate, may possibly show a yellowish coloration.

5 EP-A 47,452 describes the production of mixed trimers based on HDI and IPDI by a process wherein starting diisocyanates which are not freed from carbon dioxide are used, necessitating comparatively large quantities of catalysts, as can be seen from the
10 examples.

Other known processes for the production of isocyanurate polyisocyanates based on HDI are also attended by serious disadvantages. Thus, GB-PS 920,080, DE-OS 3,100,262, DE-OS 3,219,608 or DE-OS 3,240,613 for
15 example describe processes for the trimerization of HDI using metal-containing catalysts and co-catalysts such as phenols, alcohols or tertiary amines. The metal compounds can only be removed from the end products by very elaborate processes, if at all, and can signifi-
20 cantly affect subsequent applications and also the stability of the end products. In addition, the use of co-catalysts containing active hydrogen atoms leads to secondary reactions in which valuable isocyanate groups are consumed. The same also applies to the process
25 according to EP-AS 155,559, wherein ammonium salts of organic acids are used as catalysts in combination with large amounts of alcoholic compounds.

In the processes according to EP-A 57,653, EP-A 89,297 and EP-A 187,105, organosilicon catalysts
30 are used in comparatively large quantities. These compounds also cannot be completely removed from the end product and adversely affect its use.

Accordingly, an object of the present invention is to provide a new process for the partial trimeriza-
35 tion of the isocyanate groups of HDI which combines the following advantages:

- The end products are substantially colorless, i. ., have a color value (HAZEN) according to DIN 53,409 below 100.
- The end products are free from cloudiness and can be dissolved without cloudiness in any of the standard lacquer solvents.
- The end products contain no metal ions.
- The process can be carried out using minimal quantities of catalysts without being dependent on the use of large quantities of isocyanate-reactive co-catalysts.

It has now surprisingly been found that this object can be achieved according to the present invention as described in detail hereinafter.

SUMMARY OF THE INVENTION

The present invention is directed to a process for the production of polyisocyanates containing isocyanurate groups by trimerizing a portion of the isocyanate groups of hexamethylene diisocyanate using quaternary ammonium hydroxides as the trimerization catalyst, terminating the trimerization reaction at the desired degree of trimerization by adding a catalyst poison and/or by thermally deactivating the catalyst and removing unreacted hexamethylene diisocyanate to a residual content of at most 0.2% by weight, characterized in that

- a) the hexamethylene diisocyanate used as starting material is freed from carbon dioxide to a residual content of less than 20 ppm (weight) and
- b) the catalyst is used in a quantity of less than about 0.03% by weight, based on the weight of the hexamethylene diisocyanate used.

The present invention is also directed to the polyisocyanates containing isocyanurate groups obtained

by this process and to their use, optionally blocked by blocking agents for isocyanate groups, as the isocyanate component for the production of polyisocyanate polyaddition products.

5 DETAILED DESCRIPTION OF THE INVENTION

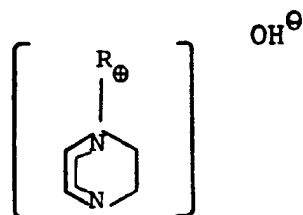
The use of HDI which is substantially free from carbon dioxide as the starting material is crucially important to the present invention. The HDI used in accordance with the invention has a carbon dioxide
10 content of less than 20 ppm (weight), preferably less than 10 ppm (weight) and more preferably less than 5 ppm (weight).

Technical HDI purified by distillation, which has previously been used for the production of poly-
15 isocyanates containing isocyanurate groups, contains considerable quantities (approximately 20 ppm to 100 ppm by weight) of carbon dioxide. Carbon dioxide can enter the HDI during the production process, for example during the phosgenation of carbonic acid salts of
20 hexamethylenediamine. It can be taken up from the air during storage and can be formed by chemical reaction of the NCO groups, for example by forming carbodiimide groups or by reaction with moisture. HDI freshly purified by vacuum distillation contains, for example,
25 40 ppm carbon dioxide after 24 hours in a sealed container. HDI stored for a period of about 6 months can contain up to 0.6% by weight carbon dioxide if the container is opened during the period of storage.

Carbon dioxide can be removed from HDI by
30 blowing ultra-pure nitrogen or a noble gas, for example argon, for example at a temperature of about 0 to 70°C, through HDI. Although it is possible to apply a higher temperature, this does not afford any significant advantages. Carbon dioxide may also be removed by
35 distillation in a stream of nitrogen or noble gas. The

method by which the carbon dioxide is removed is not crucial to the process according to the invention. However, substantially complete removal of carbon dioxide is generally not possible merely by
 5 distillation.

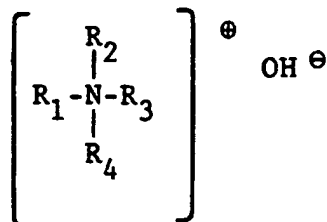
Quaternary ammonium hydroxides are used as catalyst in the process according to the invention. Basically, it is possible to use any quaternary ammonium hydroxides of the type previously recommended as
 10 trimerization catalysts for isocyanate groups. Suitable quaternary ammonium hydroxides include those according to US-PS 3,487,080 at column 2, lines 10 to 38 or according to EP-A 10,589 at page 6, line 5 to page 8, line 10 (U.S. Patent 4,324,879). Also suitable are
 15 compounds corresponding to the formula



wherein

R is an alkyl radical containing 1 to 20, preferably 1 to 4 carbon atoms, an araliphatic hydrocarbon radical containing 7 to 10, preferably 7 carbon
 20 atoms or a saturated cycloaliphatic hydrocarbon radical containing 4 to 10, preferably 5 to 6 carbon atoms.

Preferred catalysts include compounds corresponding to the formula



wherein

- R_1 , R_2 and R_3 may be the same or different and represent alkyl radicals containing 1 to 18 carbon atoms, preferably 1 to 4 carbon atoms, more preferably methyl groups and
- R_4 is a benzyl, 2-hydroxyethyl, 2-hydroxypropyl or 2-hydroxybutyl radical.

Particularly preferred catalysts include N,N,N-trimethyl-N-benzylammonium hydroxide and N,N,N-trimethyl-N-(2-hydroxypropyl)-ammonium hydroxide.

The quaternary ammonium hydroxides to be used in accordance with the invention and their production are known. They are commercially available either in the form of colorless substances or solutions or as already stated, only as brightly colored solutions. This natural color of the catalysts is often a disadvantage in the processes corresponding to the prior art cited above for the production of isocyanurate polyisocyanates based on HDI, because in these known processes the catalysts have to be used in comparatively large quantities. By contrast, in the process according to the invention, the natural color of the catalysts is not an important factor because the catalyst is only used in extremely low concentrations.

In the process according to the invention, the catalyst is used in a quantity of less than 0.03% by weight, preferably in a quantity of less than 0.01% by weight and more preferably in a quantity of from 0.0005 to 0.005% by weight, based on the HDI used. The particular optimal quantity of catalyst depends on the type of quaternary ammonium compound used and may readily be determined by a preliminary test. When N,N,N-trimethyl-N-benzyl ammonium hydroxide is used, it is sufficient to use particularly small quantities.

In contrast to the process according to JP-PS 601,337, co-catalysts, especially isocyanate-reactive co-catalysts, are not necessary and may be omitted in the process according to the invention. In particular, there is no need to use relatively large quantities of compounds containing isocyanate-reactive groups such as phenols, oximes and, in particular, methanol. Secondary reactions between a portion of the isocyanate groups of HDI and the isocyanate-reactive groups are avoided. Valuable isocyanate groups are not consumed and the formation of cloudiness attributable to these secondary products may be prevented.

Accordingly, the trimerization process according to the invention may also be carried out very effectively when no urethane groups are formed during the catalysis process. However, since many of the catalysts used in the process according to the invention are dissolved in solvents containing hydroxyl groups or themselves carry hydroxy groups, the formation of urethane groups in the process according to the invention is not ruled out. It is particularly preferred to use solvents which do not contain any isocyanate-reactive groups for the catalysts. The catalyst may also be used in solvent-free form.

When hydroxyl group-containing solvents are used, it is preferred to use those which do not form solid reaction products with HDI at room temperature and which reduce the functionality of the end products as little as possible. Hydroxyl group-containing solvents such as these include 2-ethylhexane-1,3-diol and 2-ethylhexanol. Examples of suitable solvents with no isocyanate-reactive groups include dimethylformamide, dimethylacetamide, dimethylsulfoxide and acetonitrile.

Due to the extremely small quantities of catalyst, dosing and incorporation of the pure,

undiluted catalyst, although possible, is not easy, particularly when the process is carried out continuously on an industrial scale. Accordingly, it is preferred to use heavily diluted catalyst solutions.

5 Concentrations below 5% by weight, preferably below 1% by weight are particularly suitable. To prepare catalyst solutions such as these, it is often advisable to remove solvents present in the catalysts such as methanol by mild distillation after the addition of a
10 suitable solvent of the type mentioned by way of the example above.

The trimerization reaction is preferably carried out in the absence of a solvent (apart from the solvent for the catalyst), although this does not rule
15 out the use of standard lacquer solvents during the trimerization reaction. Examples include esters such as butyl acetate or ethoxyethylacetate; ketones such as methylisobutylketone or methylethylketone; hydrocarbons such as xylene; and mixtures of such solvents. However,
20 since unreacted HDI is removed after the trimerization reaction, the use of such solvents during the trimerization reaction results in unnecessary additional expense.

To carry out the trimerization reaction, the
25 catalyst is added to HDI which has been substantially freed from carbon dioxide. The catalyst may be added in increments during the trimerization reaction. The trimerization reaction is generally carried out at a temperature of about 0 to 100°C, preferably about 20 to
30 80°C and is terminated at a degree of trimerization of about 10 to 40%, preferably about 20 to 30%. By "degree of trimerization" is meant the percentage of isocyanate groups present in the starting diisocyanate which react during the trimerization reaction.

To terminate the trimerization reaction, a suitable catalyst poison is generally added to the reaction mixture. Suitable catalyst poisons include inorganic acids such as hydrochloric acid, phosphorous acid or phosphoric acid; sulfonic acids or derivatives thereof such as methanesulfonic acid, p-toluenesulfonic acid or p-toluenesulfonic acid methyl or ethyl ester; and perfluorinated sulfonic acids such as nonafluorobutanesulfonic acid. Particularly suitable deactivators, i.e. catalyst poisons, include acidic esters of phosphorous acid or phosphoric acid such as dibutylphosphite, dibutylphosphate or di-(2-ethylhexyl)-phosphate, which are preferably used in the form of a dilute solution in HDI. The deactivators are generally added to the reaction mixture in a quantity at least equivalent to the catalyst. However, since the catalysts can partly decompose during the trimerization reaction, the addition of a sub-equivalent quantity of the deactivator is often sufficient. On the other hand, to guarantee safe termination of the reaction, it is often also advisable to use a larger than equivalent quantity, for example twice the equivalent quantity of deactivator. Accordingly, it is preferred to use deactivators (catalyst poisons) in up to twice the equivalent quantity, based on the quantity of catalyst used. When thermally labile catalysts, for example quaternary ammonium hydroxides containing hydroxyalkyl substituents at the nitrogen, are used, it is often unnecessary to add a catalyst poison. When these catalysts are used, it is often sufficient to terminate the reaction by brief heating of the reaction mixture to temperatures above 100°C (thermal decomposition, i.e. deactivation of the catalyst).

After deactivation, excess HDI is removed in a suitable manner such as extraction (for example using n-hexane as extractant) or, preferably, thin-layer distillation in a vacuum, to a residual HDI content of 5 at most 0.2% by weight, preferably less than 0.1% by weight.

The end products of the process according to the invention are colorless liquids having a color value (HAZEN) according to DIN 53,409 below 100, preferably 10 below 50, an isocyanate content of about 10 to 24% by weight and a viscosity at 23°C of about 500 to 10,000 mPa.s.

Since only very small quantities of catalyst are used in the process according to the invention, the 15 quantity of deactivator, i.e. the catalyst poison, can also be kept correspondingly small, with the result that the end products of the process according to the invention contain only very small quantities of secondary products formed from catalyst and catalyst 20 poison which remain in solution and do not affect the subsequent use of the products. Even when the process is carried out using HDI, which has not been purified beforehand in the usual way by distillation to remove traces of chlorine-containing compounds via weakly basic 25 compounds such as metal oxides or sodium hydrogen carbonate, clear and colorless end products are obtained. By virtue of their low viscosity, the end products of the process according to the invention are suitable for the production of polyisocyanate 30 polyaddition products by reaction with compounds containing at least two isocyanate-reactive groups and are particularly suitable for the production of solventless or low-solvent two-component polyurethane lacquers.

When the end products of the process according to the invention are used in accordance with the invention, they may be blocked by blocking agents for isocyanate groups. Suitable blocking agents include the compounds mentioned by way of example in EP-A 10,589, page 15, lines 14 to 26 (U.S. Patent 4,324,879).

The end products of the process according to the invention are used for the production of high-quality two-component polyurethane lacquers, preferably in combination with known polyhydroxy polyesters, polyhydroxy polyethers and, in particular, polyhydroxy polyacrylates. In addition to the relatively high molecular weight polyhydroxyl compounds mentioned, the lacquers may also contain low molecular weight, preferably aliphatic polyols. Combinations of the end products of the process according to the invention with polyhydroxyl polyacrylates represent particularly valuable two-component binders for high-quality car repair lacquers which have outstanding weather resistance.

Polyamines, particularly in blocked form as polyketimines or oxazolidines, may also be used as reactants for the end products of the process according to the invention. The quantitative ratios in which the optionally blocked polyisocyanates according to the invention and the isocyanate-reactive compounds mentioned are reacted in the production of polyisocyanate polyaddition products lacquers are selected such that for every (optionally blocked) isocyanate group, there are about 0.8 to 3, preferably about 0.9 to 1.8 hydroxyl, amino and/or carboxyl groups.

To accelerate the hardening process, it is possible to use the known catalysts from isocyanate chemistry, for example tertiary amines such as

triethylamine, pyridine, methyl pyridine, benzyldi-
methylamine, N,N-dimethylaminocyclohexane, N-methyl
piperidine, pentamethyl diethylenetriamine, N,N'-endo-
ethylene piperazine or N,N'-dimethyl piperazine; and
5 metal salts such as iron(III) chloride, zinc chloride,
zinc(II) ethylcaproate, tin(II)-2-ethylcaproate,
dibutyltin(IV) dilaurate or molybdenum glycolate.

In blocked form the products according to the
invention in combination with polyhydroxyl compounds are
10 used in particular for the production of stoving
lacquers which can be hardened at temperatures of about
80 to 180°C (depending on the blocking agent used) to
form high-quality lacquer coatings.

To prepare ready-to-use lacquers the optionally
15 blocked polyisocyanate, the polyfunctional reactant,
optionally an isocyanate polyaddition catalyst and known
additives (such as pigments, dyes, fillers and levelling
agents) are thoroughly mixed with one another and
homogenized in a standard mixing unit, such as a sand
20 mill, either with or without solvents and diluents.

The paints and coating compositions may be
applied to the article to be coated either in solution,
from the melt or in solid form by standard methods such
as spread coating, roll coating, casting, spray coating,
25 fluidized bed coating or electrostatic powder spraying.

The lacquers containing the polyisocyanates
according to the invention produce films which adhere
surprisingly well to metal substrates and which are
particularly resistant to light, color stable under heat
30 and highly abrasion-resistant. In addition, they are
distinguished by extreme hardness, elasticity, high
resistance to chemicals, high gloss, excellent weather
resistance and good pigmentability.

In the following examples, percentages are
35 percentages by weight unless otherwise indicated.

EXAMPLESEXAMPLE 1 (Preparation of catalyst solution I)

600 g 2-ethylhexane-1,3-diol were added to and stirred with 1000 g of a commercial, colorless 40% solution of N,N,N-trimethyl-N-benzylammonium hydroxide in methanol. The methanol was then removed with thorough stirring in a water jet pump vacuum at 30 to 40°C. The 40% stock solution was adjusted with additional 2-ethylhexane-1,3-diol to a catalyst concentration of about 0.5%.

EXAMPLE 2 (Preparation of catalyst solution II)

The procedure was as in Example 1, except that dimethylformamide was used instead of 2-ethylhexane-1,3-diol to replace methanol and for further dilution. A 0.5% catalyst solution in dimethylformamide was obtained.

EXAMPLE 3 (Preparation of catalyst solution III)

60 g 2-ethylhexanol were added to 100 g of a 70% solution in methanol of N,N,N-trimethyl-N-(2-hydroxypropyl)-ammonium hydroxide (prepared by the reaction of trimethylamine with propylene oxide in methanol) and the methanol was subsequently removed in a water jet pump vacuum. The solution was then adjusted with additional 2-ethylhexanol to a catalyst concentration of 4%. The solution was brown in color.

EXAMPLE 4 (According to the invention)

In a stirred reactor 3200 g HDI were degassed for about 10 minutes at about 20°C by applying a vacuum (50 mbar) and stirring vigorously. The gas space of the apparatus was then filled with pure nitrogen. A stream of pure, dry nitrogen was then vigorously passed through the liquid for about 1 hour at around 25°C. Prior to treatment the HDI had a CO₂ content of 44 ppm; the CO₂ content was reduced to 2 ppm after the described treatment. More nitrogen was passed through the reaction mixture for the remainder of the reaction.

- 32 g (0.96 mmol of base) of catalyst solution (I) were then added dropwise over a period of 15 to 30 minutes, followed by heating for 30 minutes to 60°C. Since the reaction was now slightly exothermic, the contents of the reactor were kept at 60 to 65°C by cooling. The reaction abated after about 0.5 h, at which time the NCO content of the crude product measured 42%. The crude product was then stirred for about 1 h at 60°C until an NCO content of 38% was reached. The reaction was then terminated by the addition of 0.32 g of a 25% solution of dibutylphosphate (0.38 mmol) in HDI, followed by stirring for 15 minutes. The liquid was then allowed to cool to ambient temperature and excess HDI was removed by thin-layer distillation.
- 1382 g of a clear, light polyisocyanate characterized by the following data were obtained:
Viscosity: 1800 mPa.s/23°C
Color value (HAZEN) according to DIN 53,409: 20
NCO content: 22.3%
Free HDI content: 0.05%
Dilutability with xylene: was diluted without cloudiness to below a solids content of 10%

EXAMPLE 5 (Comparison Example)

- The procedure was as in Example 4. 3200 g HDI were introduced into an apparatus which was then degassed by applying a vacuum and filled with nitrogen. Thereafter, however, no nitrogen, was blown through, so that as a result nitrogen was not introduced into the liquid but only passed over. The CO₂ content of the HDI was only negligibly reduced by this measure to 38 ppm.
- 32 g of catalyst solution (I) were then added as described in Example 1, followed by heating to 60°C. No reaction was observed and there was hardly any reduction in the NCO content. An additional 32 g of

catalyst solution were added after 4 h at 60°C; the reaction still did not start. After an additional 64 g of catalyst was added, a reaction began and was terminated at an NCO content of 38.2% by the addition of 5 1.58 g of a 25% solution of dibutylphosphate. After cooling to 25°C, the solution was in the form of a cloudy liquid which gradually precipitated a white deposit. After removal of free HDI by thin-layer distillation, a cloudy yellow product having an NCO 10 content of 21.4% was obtained. The clouding did not disappear after dilution with butyl acetate. Dilution with xylene intensified the clouding. The product was unsuitable for use in high-quality PUR lacquers.

EXAMPLE 6 (According to the invention)

15 In a stirred reactor, 798 g of freshly distilled hexamethylene diisocyanate were vigorously stirred in a vacuum (<50 mbar) for 30 minutes at 20°C. The gas space of the apparatus was then filled with highly pure nitrogen. The carbon dioxide content of the 20 HDI was 44 ppm. A stream of pure, dry nitrogen was then vigorously passed through the liquid for 1 hour at 30 to 40°C. Re-determination of the CO₂ revealed a content of 2 ppm.

Throughout the reaction, dry nitrogen was 25 passed through the reaction mixture. To initiate the trimerization reaction, 12 g catalyst solution II were added dropwise over a period of about 30 minutes, followed by slow heating to 70°C. The reaction was exothermic and was sustained for 1 hour at about 75°C 30 without any further supply of heat. Another 12 g of catalyst solution were then added. The reaction mixture was then left to react for another 30 minutes with thorough stirring. The reaction was terminated at an NCO content of 42.4% by the addition of 0.6 g 35 (equivalent ratio of catalyst to terminator = approx.

1:1) of a 25% solution of dibutylphosphate in HDI.
After 15 minutes, the reaction mixture was subjected to
thin-layer distillation at 130°C to separate solvent and
HDI. 230 g of a polyisocyanate characterized by the
5 following data were obtained:
Viscosity: 2200 mPa.s/23°C
NCO content: 22.0%
Free HDI content: 0.09%
Color value: 30 (DIN 53,409)

10 EXAMPLE 7 (According to the invention)

The procedure was as in Example 6. The
reaction was terminated at an NCO content of 38.0%.
After termination of the reaction and after thin-layer
distillation at 120°C, 350 g of a product having the
15 following characteristic data were obtained:
Viscosity: 3000 mPa.s/23°C
NCO content: 21.7%
Color value: 40 (DIN 53,409)
Free HDI content: 0.1%

20 EXAMPLES 8-12 (According to the invention)

The procedure was as described in Example 1,
i.e. CO₂ was removed from the HDI with a vigorous stream
of nitrogen at 40 to 50°C. The other conditions of the
polymerization reaction are shown in Table 1. The
25 catalyst was deactivated with dibutylphosphate with one
exception (Example 10). Table 1 also sets forth the NCO
content of the reaction mixture at which termination of
the polymerization reaction was initiated. Table 2 sets
forth the characteristic data of the end product after
30 removal of excess HDI.

TABLE 1

EXAMPLE	HDI (g)	CO ₂ CONTENT AFTER N ₂ TREATMENT	CATALYST SOLUTION OF EXAMPLE 3	DEACTIVATION BY HEATING OR WITH DIBUTYLPHOSPHATE (25% SOLUTION IN HDI)	NCO CONTENT OF THE CRUDE SOLUTION
8	3000	4 ppm	12.4 g	2 g at 50°C	39.2 %
9	3000	6 ppm	12.4 g	0.5 g at 60°C	34.5 %
10	2500	2 ppm	7 g	by heating to 120°C	41.3 %
11	2500	2 ppm	-	0.13 g at 60°C	42.0 %
12	2500	10 ppm	-	0.52 g at 70°C	38.1 %

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TABLE 2

EXAMPLE	QUANTITY OF PRODUCT AFTER REMOVAL OF HDI	NCO CONTENT	VISCOSITY 23°C /mPa.s	COLOR VALUE (HAZEN), DIN 53,409	FREE HDI CONTENT
8	1296 g	22.8 %	1700	70	0.12 %
9	560 g	20.2 %	9800	90	0.09 %
10	880 g	23.2 %	1800	70	0.05 %
11	805 g	23.7 %	1500	30	0.05 %
12	1200 g	22.0 %	2100	40	0.05 %

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Although the invention has been described in detail in the foregoing for the purpose of illustration, it is to be understood that such detail is solely for that purpose and that variations can be made therein by those skilled in the art without departing from the spirit and scope of the invention except as it may be limited by the claims.

The embodiments of the invention in which exclusive property or privilege is claimed are defined as follows:

1. A process for the preparation of a polyisocyanate containing isocyanurate groups which comprises
 - a) trimerizing a portion of the isocyanate groups of hexamethylene diisocyanate, which contains less than 10 ppm by weight of carbon dioxide, in the presence of a positive amount of less than 0.03% by weight, based on the weight of hexamethylene diisocyanate, of a quaternary ammonium hydroxide catalyst,
 - b) terminating the trimerization reaction at the desired degree of trimerization by the addition of a catalyst poison and/or by thermal deactivation and
 - c) subsequently removing unreacted hexamethylene diisocyanate such that said polyisocyanate contains at most 0.2% by weight of hexamethylene diisocyanate, based on the weight of said polyisocyanate.
2. The process of Claim 1 wherein said catalyst is used in a positive amount of less than 0.01% by weight, based on the weight of hexamethylene diisocyanate.
3. The process of Claim 1 wherein said catalyst comprises N,N,N-trimethyl-N-benzyl-ammonium hydroxide.
4. The process of Claim 1 wherein said catalyst comprises N,N,N-trimethyl-N-(2-hydroxyethyl)-ammonium hydroxide.
5. The process of Claim 1 wherein said catalyst comprises N,N,N-trimethyl-N-(2-hydroxypropyl)-ammonium hydroxide.

6. The process of Claim 1 wherein the trimerization reaction is terminated with a catalyst poison comprising dibutyl phosphate.

5 7. The process of Claim 1 wherein the trimerization reaction is terminated with a catalyst poison comprising di-(2-ethylhexyl)-phosphate.

8. A process for the preparation of a polyisocyanate polyaddition product which comprises

- 10 a) preparing a polyisocyanate based on hexamethylene diisocyanate and containing isocyanurate groups in accordance with the process of Claim 1 and
- 15 b) reacting the product of step (a) with a compound containing at least two isocyanate-reactive groups.

9. The process of Claim 8 wherein said polyisocyanate polyaddition product is a polyurethane and said compound containing at least two isocyanate-reactive groups comprises a polyol.



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A PROCESS FOR THE PRODUCTION OF POLYISOCYANATES
CONTAINING ISOCYANURATE GROUPS AND THEIR USE

ABSTRACT OF THE DISCLOSURE

The present invention is directed to a process for the production of polyisocyanates containing isocyanurate groups by trimerizing a portion of the isocyanate groups of hexamethylene diisocyanate using quaternary ammonium hydroxides as the trimerization catalyst, terminating the trimerization reaction at the desired degree of trimerization by adding a catalyst poison and/or by thermally deactivating the catalyst and removing unreacted hexamethylene diisocyanate to a residual content of at most 0.2% by weight, characterized in that

- a) the hexamethylene diisocyanate used as starting material is freed from carbon dioxide to a residual content of less than 20 ppm (weight) and
- b) the catalyst is used in a quantity of less than about 0.03% by weight, based on the weight of the hexamethylene diisocyanate used.

The present invention is also directed to the polyisocyanates containing isocyanurate groups obtained by this process and to their use, optionally blocked by blocking agents for isocyanate groups, as the isocyanate component for the production of polyisocyanate polyaddition products.

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